Modeling of the shrinkage kinetics of coffee berries during drying

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ABSTRACT. The effects of drying conditions on the shrinkage of Coffea arabica L., cultivar Catuaí Vermelho, were studied. An exponential model derived from the kinetic model was used to describe the shrinkage kinetics. In addition, an equation was proposed to describe the relationship between the superficial areas and weights of the coffee berries. Coffee berries were dried at temperatures (T) of 35, 45, 55 and 65°C and at relative humidities (φ) of 25, 35, 45 and 55%, forming 16 different sets of conditions, with an initial moisture content for all conditions of 2.27 dry basis (d.b.), dried until 0.11 d.b. Changes in the product dimensions were measured continuously during drying. Surface area-to-volume ratio values increased with moisture content decrease. The model was suitable for representing the shrinkage kinetics, as well the empirical equation for predicting surface area. Temperature had a greater influence over the constant rate of change of the characteristic dimension (k values) than φ. Activation energies were also determined to have magnitudes of 39.31, 37.32, 36.28 and 36.22 kJ mol⁻¹ for relative humidities of 25, 35, 45 and 55%, respectively.

Keywords: surface area-to-volume ratio, post-harvest, activation energy, Coffea arabica L, shrinkage.

RESUMO. Modelagem da cinética de contração de frutos de café durante a secagem. Neste trabalho foi estudado o efeito das condições de secagem sobre a contração de frutos de café (Coffea arabica L.), cultivar Catuaí Vermelho. Um modelo exponencial derivado do modelo cinético foi utilizado para descrever a cinética da contração. Ademais, uma equação foi proposta para descrever a relação entre a área superficial e a massa dos frutos de café. Frutos de café foram secos às temperaturas de (T) de 35, 45, 55 e 65°C e a umidades relativas (φ) de 25, 35, 45 e 55%, totalizando 16 condições diferentes, com um teor de água inicial de 2.27 base seca (b.s.), secos até 0.11 b.s. As alterações nas dimensões do produto foram mensuradas continuamente durante a secagem. Os valores de relação superfície-volume aumentaram com a diminuição do teor de água. O modelo foi adequado para representar a cinética da contração, bem como a equação empírica para predição da área superficial. A temperatura teve maior influência sobre a taxa constante de mudança da dimensão (k) que φ. Energias de ativação foram obtidas, com magnitudes de 39,31; 37,32; 36,28 e 36,22 kJ mol⁻¹ para umidades relativas de 25, 35, 45 e 55%, respectivamente.


Introduction

The drying or dehydration of foods is one of the most common methods for assuring the quality and stability of agricultural products, decreasing the water activity of the material, reducing the biological activity, and reducing the chemical and physical changes of the material during storage. Moisture loss leads to modifications in cellular structure, consequently altering product shape and dimension. This information is particularly important to coffee because it presents a high level of moisture content during the initial stages of drying. This moisture loss, according to Yan et al. (2007), influence the products’ physical properties and modify the final texture and transport properties of the dry foods.

The volumetric behavior of a grain or fruit, during the drying process, is important to estimate tension levels that the product is submitted, as well when developing equipment, processes and simulation models. This behavior is dependent of temperature and humidity gradients. Additionally,
the superficial area, shape and dimensions of an agricultural product are crucial data for mass and heat transfer studies as well as for airflow within a granular mass of the product. These parameters and additional information, such as moisture content, are used for predicting storage and drying conditions of grain and cereals and quality losses during the post-harvest period.

Agricultural products with high initial moisture contents such as coffee (approx. 60% w.b.) (RESENDE et al., 2009) easily lose moisture via diffusion to the surrounding environment, due to the great water vapor pressure difference between the product and the air. Depending upon the evaporative demand of drying air, the rate of moisture loss contributes to a marked reduction of the moisture content, material dimensions and, consequently, volume.

Shrinkage is the ratio between the initial volume and the volume at a certain time after the moisture loss processes. According to Ramos et al. (2005), this trend is due to the reduction of the product size (macroscopic phenomenon), which is a reflection of the reduction of its cellular dimensions (microscopic phenomena) because of the loss of moisture. Aguilera (2004) reported that shrinkage during the dehydration of agricultural products occurs when the viscoelastic matrix contracts into the space previously occupied by the water removed from the cells.

Bennamoun and Belhamri (2006), Ramos et al. (2005) and Mayor and Sereno (2004) reported that shrinkage is an important phenomenon that is frequently neglected during the development of drying models. Neglect of this phenomenon leads to inefficient machinery and handling. The correct quantification of this trend is vital because it allows for the analysis of drying kinetics and, consequently, of the whole drying process (HASHEMI et al., 2009).

Shrinkage has been correlated linearly to moisture content (DISSA et al., 2008; BATISTA et al., 2007). Hatamipour and Mowl (2002) reported a linear correlation for the volume change in carrots during the drying process. Ramos et al. (1994) studied microstructural changes in grape cells during the first stage of convective air drying and concluded that even though cell dimensions suffered modifications during drying, the cell shapes remained unchanged.

Several authors have studied the influence of different process conditions on the volume change of materials during dehydration. McMin and Magee (1997) analyzed the effect of temperature on the drying of potatoes; Ratti (1994) investigated the influence of air velocity and relative humidity on the drying of potatoes, apples and carrots; and Afonso Júnior et al. (2003) examined the effect of varying moisture content on the reduction of superficial area, volume and equivalent sphere diameter of coffee berries. However, these studies have not satisfactorily concluded the different conditions that influence the process of shrinkage.

The aim of this work was to determine the influences of various drying temperatures and relative humidities on the shrinkage of coffee berries during the drying process. These include changes in area and volume, which directly affect the visual attributes of the fruit. In addition to the above-mentioned parameters, a new correlation between the surface area and weight of coffee berries was investigated.

Material and methods

The present work was conducted in the Laboratory of Physical Properties and Quality Evaluation of Agricultural Products at the National Grain Storage Training Center – CENTREINAR, Federal University of Viçosa, Viçosa, Minas Gerais State, Brazil.

Coffee berries (Coffea arabica L.) from Catuaí Vermelho cultivar were purchased from a local farmer. They were manually harvested at the cherry stage with an initial moisture content of 2.27 d.b. Immature, deteriorated or damaged fruits were eliminated, with the purpose of obtaining a more homogeneous material with elevated quality.

Initial moisture contents were achieved by drying the product in an oven at 105 ± 1°C, until a constant mass was obtained. These were performed in triplicate. The reduction of moisture content over drying time was supervised through weight loss until a final moisture content of 0.11 d.b was reached. To monitor weight loss, a digital chronometer and an analytical balance with a precision of 0.01 were utilized.

The drying treatments were placed in a 4 x 4 factorial scheme, with four temperature levels (35, 45, 55 and 65°C) and four levels of relative humidities of drying air (25, 35, 45 and 55%) in a completely randomized design. The treatments conditions were achieved using a temperature-controlled chamber, manufactured by Aminco, model Aminco-Aire 150/300 CFM. Removable perforated trays were placed inside the equipment to allow air to pass through the samples, each of which contained 50 g of product. Airflow was monitored with an anemometer with rotating blades and kept at around 10 m3 min⁻¹ m⁻². Temperature and air relative humidity were monitored with a
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psychrometer installed next to the trays containing the samples.

Product dimensions were measured across the three orthogonal axes of the coffee berries using digital calipers after weighing the samples over a moisture content range of 2.27 to 0.11 d.b., with ten repetitions for each condition, in order to obtain product volumes \( V \).

### Shrinkage

The association between physical processes and time can be described according to the classic kinetic model (Equation 1), in which \( P \) represents the physical characteristic being evaluated (in this case, volume); \( k \) is the constant rate of change of the characteristic dimension (h\(^{-1}\)) that correlates volume and time, dependent on the environment conditions in which the process occurs (LESPINARD et al., 2009); and \( m \) is the reaction order degree, which is dimensionless.

\[
\frac{dP}{dt} = kP^m \quad (1)
\]

For a various number of physical processes, it is difficult to distinguish between zero, first and second order kinetics. This fact can be explained by the small extension of reaction and the small evaluation rate of the studied parameter. Therefore, the \( m \) value was considered to be equal to one (first order) to facilitate the calculation procedure, and to linearly analyze the shrinkage phenomenon.

Data was evaluated using a non-linear regression analysis performed via STATISTICA 7.0\(^{®} \) software using an exponential model (Equation 2), which was obtained through Equation 1 after mathematical rearrangements.

\[
\exp \left( \frac{V_t - V_0}{V_0 - V_e} \right)^{1/m} = \exp \quad (2)
\]

where:
- \( V_0 \): initial volume (cm\(^3\));
- \( V_t \): volume at a certain time (cm\(^3\));
- \( V_e \): equilibrium volume (cm\(^3\));
- \( t \): time (h).

The equilibrium volume was obtained from the average of the last three volumes. This was adopted since these values were practically constant, indicating that the equilibrium had been reached.

The best fit model was evaluated via calculation of the mean relative error (MRE) and the standard error estimation (SEE), Equations 3 and 4, respectively. The determination coefficient (R\(^2\)) was also utilized.

\[
\text{MRE} = 100 \sum_{i=1}^{n} \left( \frac{X_{exp} - X_{pre}}{X_{exp}} \right) \quad (3)
\]

\[
\text{SEE} = \sqrt{\frac{\sum_{i=1}^{n}(X_{exp} - X_{pre})^2}{DF}} \quad (4)
\]

where:
- \( n \): number of observations, dimensionless;
- \( X_{exp} \): experimental values of the equilibrium moisture content (d.b.);
- \( X_{pre} \): calculated values of the equilibrium moisture content (d.b.);
- \( DF \): degrees of freedom.

For practical purposes, values below 10% of the mean relative error indicate a good fit (MOHAPATRA; RAO, 2005).

### Activation Energy

Activation energy is defined as the minimum energy necessary to initiate a chemical reaction. In the drying process, the lower the activation energy is, the higher the water diffusivity in the product will be. This means that lower values of activation energy allow for a higher moisture transfer rate and, consequently, lower drying time and cost.

Arrhenius Equation (Equation 5) is widely used to model and evaluate parameters that modify with temperature, including shrinkage, during the drying process of agricultural products. This equation, in addition with STATISTICA 7.0\(^{®} \) software, was employed to obtain the values of activation energy of the cultivar studied in this work. This equation states the relationship between activation energy and the velocity at which the reaction occurs.

\[
k = A_0 \exp \left( \frac{-E_a}{RT_{atm}} \right) \quad (5)
\]

where:
- \( A_0 \): pre-exponential factor (h\(^{-1}\));
- \( E_a \): activation energy (kJ mol\(^{-1}\));
- \( R \): universal gas constant (8.314 kJ mol\(^{-1}\) K\(^{-1}\));
- \( T_{atm} \): temperature (K).

### Surface area-to-volume ratio (SVR)

SVR was obtained from Equations 6 and 7. It was considered that the coffee berry could be represented by a prolate spheroid shape type (MOHSENIN, 1986), as seen in Figure 1.
\[ V = \frac{\pi abc}{6} \]  \hspace{1cm} (6)

\[ S = \frac{\pi b^2}{2} + \frac{\pi ab}{2e} \sin^{-1} e \]  \hspace{1cm} (7)

\[ e = \left[1 - \left(\frac{b}{a}\right)^2\right]^{1/2} \]  \hspace{1cm} (8)

where:

- \( V \): volume (cm\(^3\))
- \( S \): surface area (cm\(^2\))
- \( a \): longest intercept (mm)
- \( b \): longest intercept normal to \( a \) (mm)
- \( c \): longest intercept normal to \( a \) and \( b \) (mm)
- \( e \): eccentricity (dimensionless)

**Figure 1.** Schematic draw of a coffee berry considered as a prolate spheroid shape type and its characteristics dimensions.

**Surface area (S) and weight (W) relationship**

Mohsenin (1986) states a relationship between weight and surface area for spheroid products (Equation 9):

\[ S = \xi W^d \]  \hspace{1cm} (9)

where:

- \( S \): surface area, cm\(^2\)
- \( d, \xi \): constant, dimensionless
- \( W \): weight (g)

Quantifying the superficial area of agricultural products is a difficult process due to the fact that these foodstuffs do not have a uniform and homogeneous form. As a result, a great amount of time and many replicates are required, resulting in a large range of errors. On the other hand, weight measurement is easy to obtain with a simple weighing scale; therefore this equation provides a simple approach for acquiring values of superficial area by means of product weight. This information is essential for developing and improving drying equipment and helping studies of mass and heat transfer.

**Results and discussion**

**Shrinkage Evaluation**

The constant rate values \((k)\) for Catuái Vermelho cultivar in each \(T\) and \(\phi\) combination are shown in Table 1, in which \(k\) values were analyzed varying along with \(T\) and \(\phi\) (Figure 2) in order to assess which factor had the highest influence over \(k\).

Equation 10 represents the surface shown in Figure 2. The modeling procedure was carried out by means of the software Table Curve 3D® v.4.0, which provided a \(R^2\) of 0.9996.

\[ k = -0.0931346875 - 0.0008935 T + 0.7045 RH + 0.000118 T^2 - 0.50625 RH^2 - 0.00973 T \cdot RH \]  \hspace{1cm} (10)

The figure presented allow us to conclude that \(k\) is more dependent on temperature variability, and during the drying process, \(k\) values decrease with higher \(\phi\) values and increase with rises in temperature. These results are supported by other researchers. Krokida et al. (2000) considered that the temperature is the parameter with the most influence over food drying kinetics, and Afonso Júnior et al. (2004) also observed this trend for coffee berries. Ramos et al. (1994) found that temperature increased the rate of cellular shrinkage, with an Arrhenius type behavior.

Figure 3 demonstrates the shrinkage phenomenon of coffee berries at the beginning and the end of the drying process. Ten repetitions were made for each berry continuously during drying, and an average of \(a\),
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and c values were attained in order to obtain the product volume. Volume values ranged from 10.59 x 10^-7 to 6.43 x 10^-7 m at the beginning and end of drying, respectively. The total shrinkage was 1.22, 2.23 and 2.18 mm for a, b and c product dimension, respectively.

Figure 3. Total shrinkage representation of coffee berries.

Shrinkage is more relevant at the initial stages of drying. As the drying process occurs, the product tends toward an equilibrium state. As such, it starts to become more difficult for shrinkage to occur, leading to a stability of the product dimension characteristics. In addition, a combination of T and φ that provides a higher drying air potential (high T and low φ) also contributes to a stronger constant rate of change of the characteristic dimension (higher k) during dehydration. However, the product dimensions at equilibrium do not alter with air characteristics, but rather solely with the time to reach the equilibrium, which is diminished.

Activation energy

Activation energies obtained were 39.31, 37.32, 36.28 and 36.22 kJ mol^-1 for relative humidities of 25, 35, 45 and 55%, respectively. Corrêa et al. (2006) also obtained the activation energy for coffee fruits; however, due to different cultivars and drying conditions used, their value (22.6 kJ mol^-1) differs from the present study, denoting the importance of evaluating different agricultural products and cultivars.

Figure 4 illustrates the activation energies found in this study. Examining this figure, it can be noticed that at higher φ (45-55%), the activation energy practically remained constant, marked by a breaking point at 45% φ. Furthermore, $E_a$ increased with a decrease in φ. These features mean that at superior φ importance, it is easier to remove superficial moisture and less energy is required to do so. This trend can be explained by the following: higher φ values leads to lower water vapor pressure difference between the product and the drying air and consequently higher moisture content at the product surface (more water availability). Therefore, it becomes more difficult to remove moisture as the drying process occurs. Therefore, the highest k values are expected for a drying air with the highest potential for decreasing the moisture content of the product (lower φ and higher T), as seen in Table 1.

Table 1. Estimated constant rates of change of characteristic dimension (k) for Catuaí Vermelho cultivar.

<table>
<thead>
<tr>
<th>Temperature (ºC)</th>
<th>φ (%)</th>
<th>35</th>
<th>45</th>
<th>55</th>
<th>65</th>
</tr>
</thead>
<tbody>
<tr>
<td>25</td>
<td>0.0826</td>
<td>0.1397</td>
<td>0.2231</td>
<td>0.3338</td>
<td></td>
</tr>
<tr>
<td>35</td>
<td>0.0843</td>
<td>0.1362</td>
<td>0.2116</td>
<td>0.3126</td>
<td></td>
</tr>
<tr>
<td>45</td>
<td>0.0791</td>
<td>0.1236</td>
<td>0.1890</td>
<td>0.2785</td>
<td></td>
</tr>
<tr>
<td>55</td>
<td>0.0670</td>
<td>0.1018</td>
<td>0.1550</td>
<td>0.2309</td>
<td></td>
</tr>
</tbody>
</table>

Surface area-to-volume ratio (SVR)

Figure 5 demonstrates that the SVR tends to decrease with an increase in moisture content (d.b.).

As the moisture content increases, both the surface area and volume increase. However, the proportion of the volume increase is higher than that of the surface area, leading to lower SVR values.

Relationship between surface area (S) and weight (W)

The ξ and d parameters were obtained through STATISTICA 7.0® software and provided MRE, SEE
and $R^2$ values of 2.077, 0.1392 and 0.9134, respectively. This indicates that the data fit well to the empirical equation, allowing the use of Equation 11 for predicting surface area using the coffee berry weight, at a moisture content range of 0.11 to 2.27 d.b.

$$S = 3.982756W^{0.278892}$$

(11)

Conclusion

The shrinkage phenomenon can be represented through the model here suggested for the Catuá Vermelho cultivar of coffee berries at the temperatures and relative humidity ranges of 35 to 65°C and 25 to 55%, respectively. SVR had the tendency to decrease as moisture content increased, and a prediction of surface area using the weight of the coffee berry was satisfactory with respect to the empirical equation formulated in the moisture content range studied in this work. Constant rates of change of the characteristic dimension are more dependent on temperature variability than $\phi$.

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References


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